

the photographers'
FORMULARY, INC.
Chemical and Laboratory Resources

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December 17, 1978

Miss Laura Gilpin
409 Camino del Monte Sol
Santa Fe,
New Mexico 87501

Dear Miss Gilpin:

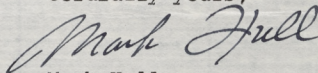
You may recall that some months ago we sent to you a small sample of ferric oxalate for you to try in your printing. You found it to dissolve with difficulty and to produce weak platinumums.

We since have begun to make our own ferric oxalate, and would be grateful if you would try it as you did the earlier product. The material we now have is in liquid form, and should be very close to the strength you are accustomed to mixing. It does not contain any potassium chlorate, so you may want to add some.

The small bottle we are sending you contains 50 ml of what should be 20% ferric oxalate with a slight excess of oxalic acid. It is quite sensitive to light and heat, and we have found that the glass bottle is not sufficient to protect it from light.

We would be most appreciative for any suggestions you might want to make after using the ferric oxalate, and look forward to hearing from you.

Cordially yours,



Mark Hull
Photographers' Formulary, Inc.

P.S. Merry Christmas!